

Characteristics of MW

Spectra

(1) It is caused by rotation of molecules around its centre of gravity.

(2) Small amount of energy change is involved.

$$E = 10 \text{ to } 10^3 \text{ joule/mole}$$

$$\bar{\nu} = 1 \text{ to } 10^2 \text{ cm}^{-1}$$

(3) Spectral range \rightarrow MW

(4) It is characterized by rotational quantum no. (J)

$$J = 0, 1, 2, 3, \dots$$

(5) For rotational energy level

$$E_{\text{rot}} = \bar{B} h c J(J+1) \text{ ergs}$$

cmol

$$\bar{\nu} = \bar{B} J(J+1) \text{ cm}^{-1}$$

(6) Frequency separation between rotational ~~lines~~ energy level

$$\Delta E_{rot} = 2Bhc(j+1)$$

and

$$\Delta j = 2B(j+1)$$

i.e. $2B, 4B, 6B, 8B$

(1) Rotational energy levels are equally populated as energy difference is small

(2) Probability of transitions are almost the same.

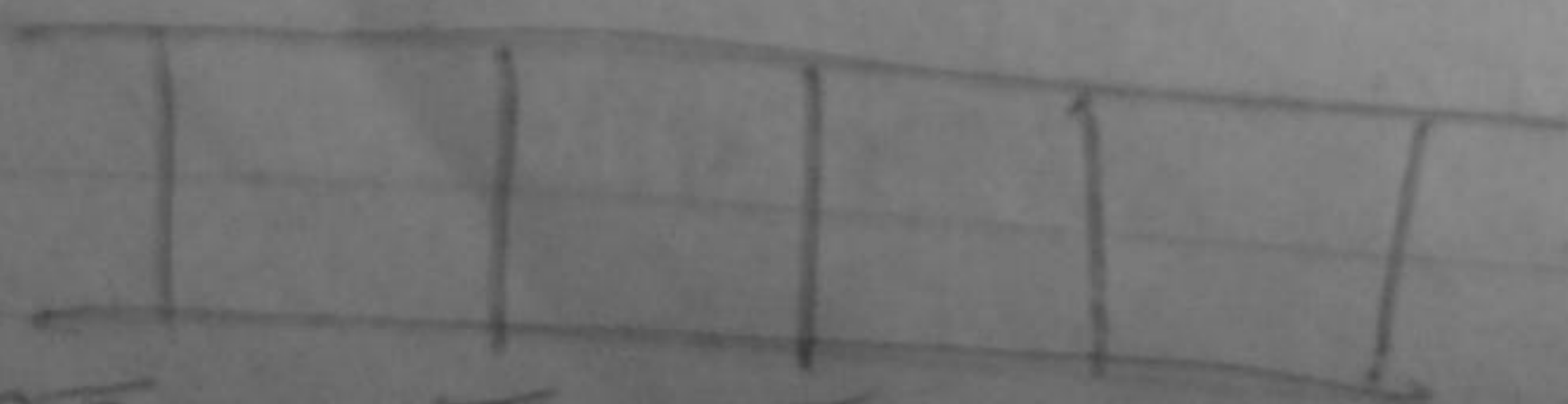
(3) Selection rule -

$$(a) \Delta j = \pm 1$$

(b) molecule has permanent dipole moment

(4) Frequency of rotational lines

$$2B, 4B, 6B, 8B$$



(11) Rotational lines are equally spaced and $\Delta \bar{\nu} = 2B$ cm^{-1} .

(12) Only molecules with permanent dipole moment are IR active. Orientation of dipole moment changes as the molecule rotates and as a result electric component of dipole moment fluctuates.

(13) IR is studied in gas phase only.

(14) Distortion of bond length occurs at J values equal to or greater than 10. ($J \geq 10$)

(15) Bond length, bond angle, no. of ν , isotopic weight etc. can be calculated.